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APPLICATION NO.	F	ILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.	
10/512,046	10/512,046 10/21/2004		Shojiro Tanase	HEIW:040	8407	
27890	7590	07/01/2005		EXAMINER		
STEPTOE & JOHNSON LLP				CHOI, LING SIU		
1330 CONNECTICUT AVENUE, N.W. WASHINGTON, DC 20036			ART UNIT	PAPER NUMBER		
, , , ,				1713		

DATE MAILED: 07/01/2005

Please find below and/or attached an Office communication concerning this application or proceeding.

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		Application No.	Applicant(s)	
		10/512,046	TANASE ET AL.	•
	Office Action Summary	Examiner	Art Unit	
		Ling-Siu Choi	1713	
Period fo	The MAILING DATE of this communic or Reply	ation appears on the cover	sheet with the correspondence a	ddress
THE   - External after - If the   - If NC   - Failu   Any I	ORTENED STATUTORY PERIOD FO MAILING DATE OF THIS COMMUNIC nsions of time may be available under the provisions of SIX (6) MONTHS from the mailing date of this communic period for reply specified above is less than thirty (30) operiod for reply is specified above, the maximum stature to reply within the set or extended period for reply we reply received by the Office later than three months after the patent term adjustment. See 37 CFR 1.704(b).	ATION. 37 CFR 1.136(a). In no event, howe nication. days, a reply within the statutory min tory period will apply and will expire still, by statute, cause the application to	ver, may a reply be timely filed imum of thirty (30) days will be considered time SIX (6) MONTHS from the mailing date of this of become ABANDONED (35 U.S.C. § 133).	
Status				
1)	Responsive to communication(s) filed	on .		
· —	•	b)⊠ This action is non-fina	al.	
,	Since this application is in condition for closed in accordance with the practice	•	· ·	e merits is
Dispositi	ion of Claims			
5)□ 6)⊠ 7)□	Claim(s) 1-12 is/are pending in the ap 4a) Of the above claim(s) is/are Claim(s) is/are allowed.  Claim(s) 1-12 is/are rejected.  Claim(s) is/are objected to.  Claim(s) are subject to restriction	withdrawn from considera		
Applicati	on Papers			•
10)⊠	The specification is objected to by the The drawing(s) filed on <u>21 October 200</u> Applicant may not request that any objection Replacement drawing sheet(s) including the oath or declaration is objected to be	04 is/are: a)  accepted on to the drawing(s) be held ne correction is required if the	in abeyance. See 37 CFR 1.85(a). e drawing(s) is objected to. See 37 C	FR 1.121(d).
	•	,		
12) a)[	Acknowledgment is made of a claim for All b) Some * c) None of:  1. Certified copies of the priority do  2. Certified copies of the priority do  3. Copies of the certified copies of application from the International See the attached detailed Office action	ocuments have been rece ocuments have been rece the priority documents ha al Bureau (PCT Rule 17.2)	ived. ived in Application No ve been received in this National (a)).	I Stage
Attachment	t(s)			
1) Notic 2) Notic 3) Inform	e of References Cited (PTO-892) e of Draftsperson's Patent Drawing Review (PTo- nation Disclosure Statement(s) (PTO-1449 or P' r No(s)/Mail Date 10/21/2004.	D-948) FO/SB/08) 5) 🔲 (	Interview Summary (PTO-413) Paper No(s)/Mail Date Notice of Informal Patent Application (PTO) Other:	O-152)

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## **DETAILED ACTION**

1. This Application is a 371 of PCT/JP03/05615 filed May 2, 2003. Claims 1-12 are now pending, wherein claims 1-10 are drawn to a solid catalyst component; claim 11 is drawn to a catalyst; and claim 12 is drawn to a method to produce an olefin polymer.

2. JP 58-811 cited in Form PTO 1449 has not be found. Thus, it will not be considered in this Office Action.

## Claim Rejections - 35 USC § 103

- 3. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
  - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 4. Claims 1-12 are rejected under 35 U.S.C. 103(a) as being unpatentable over Tanase et al. (EP 1 108 730 A1) in view of Morini et al. (WO 98/56830).

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	A solid catalyst component for olefin polymerization obtained by reacting the following compounds a-b-d or a-b-c-d					
а	a halogen-containing titanium compound					
b	an alkoxylated magnesium compound obtained by reacting metal magnesium, an alcohol, and a halogen and/or halogen-containing compound containing at least					
	0,0001 gm atom of a halogen atom per mole of the metal magnesium					
С	a halogen-containing silicon compound					
d	electron-donating compound(s) represented by formula I and/or formula II,					
	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$					

(summary of claim 1)

Tanase et al. disclose a catalyst for olefin polymerization, comprising (A) a solids catalyst component prepared by contacting a magnesium compound, a titanium compound of  $Ti(OR)_nX_{4-n}$  with n=0-4, and an electron donor, wherein the magnesium compound is obtained by reacting a metal magnesium, an alcohol, and at least 0.0001 gm atoms of halogen per gm atom of magnesium, (B) an organometallic compound, and (C) an electron donor (abstract; [0004]).

The difference between the present claims and the disclosure of Tanasde et al. is the requirement of the specific poly(di)ether and/or ester of malonic acid to be used in the present invention.

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It is noted that Tanase et al. do recognize polyether (diether) or ester of malonic acid among a list of the internal electron donors to be used as an internal electron donor ([0027]; [0028]). Morini et al. disclose that the use of the specific ester of maolonic acid as an internal electron donor leads to an excellent balance between polymerization yield and isotactic index for the resulting polymer (page 3, lines 7-9). In light of such benefit, it would have been obvious to one of ordinary skill in the art at the time the invention was made to use the specific ester of malonic acid in the disclosure of Tanase and thereby obtain the present invention.

5. Claims 1-12 are rejected under 35 U.S.C. 103(a) as being unpatentable over Koyama et al. (US 5,177,043) in view of Morini et al. (WO 98/56830).

Koyama et al. disclose a catalyst for olefin polymerization, comprising (A) a solid component comprising (I) a magnesium compound obtained by contacting metallic magnesium, a halogenated hydrocarbon, and an alkoxy group-containing compound, (II) a titanium compound such as titanium tetrachloride, and (III) an internal electron donor, (B) an organoaluminum cocatalyst, and (C) a silane compound (abstract; col. 3, lines 9-12; col. 4, lines 6-18; Example 1; claim 1).

The difference between the present claims and the disclosure of Koyama et al. is the requirement of the specific poly(di)ether and/or ester of malonic acid to be used in the present invention.

It is noted that Koyama et al. do recognize diethylmalonate or diisobutyl malonate among a list of the internal electron donors to be used as an internal electron

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donor (col. 4, lines 56-57). Morini et al. disclose that the use of the specific ester of maolonic acid as an internal electron donor leads to an excellent balance between polymerization yield and isotactic index for the resulting polymer (page 3, lines 7-9). In light of such benefit, it would have been obvious to one of ordinary skill in the art at the time the invention was made to use the specific ester of malonic acid in the disclosure of Koyama et al. and thereby obtain the present invention.

6. Claims 1-12 are rejected under 35 U.S.C. 103(a) as being unpatentable over Shinozaki et al. (US 6,537,942 B2) in view of Morini et al. (WO 98/56830).

Shinozaki et al. disclose a catalyst for olefin polymerization, comprising (I) a contact product obtained by contacting (A) a solid titanium catalyst component, (B) an organometallic compound, and (C) a specific polyether compound, (II) a specific organosilicon compound, and (III) an organometallic compound, wherein the component A comprises titanium tetrahalide and alkoxymagnesium chloride (abstract; col. 9, lines 1-3; col. 10, lines 7-9).

The difference between the present claims and the disclosure of Shinozaki et al. is the requirement of the specific poly(di)ether and/or ester of malonic acid to be used in the present invention.

It is noted that Shinozaki et al. do recognize diisobutyl methylmalonate among a list of the internal electron donors to be used as an internal electron donor (col. 10, line 52). Morini et al. disclose that the use of the specific ester of maolonic acid as an internal electron donor leads to an excellent balance between polymerization yield and

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thereby obtain the present invention.

isotactic index for the resulting polymer (page 3, lines 7-9). In light of such benefit, it would have been obvious to one of ordinary skill in the art at the time the invention was made to use the specific ester of malonic acid in the disclosure of Shinozaki et al. and

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Conclusion

7. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Ling-Siu Choi whose telephone number is 571-272-

1098.

If attempt to reach the examiner by telephone are unsuccessful, the examiner=s

supervisor, David Wu, can be reach on 571-272-1114.

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LING-SUI CHOI PRIMARY EXAMINER

June 15, 2005